

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597286>

### A Novel Approach to Khantiokerically Pure Carbocyclic Nucleoside Analogues

Michael Bodenteich<sup>a</sup>; Kurt Faber<sup>a</sup>; Gerhard Penn<sup>a</sup>; Herfried Griengl<sup>a</sup>

<sup>a</sup> Institute of Organic Chemistry, Technical University Graz, Graz, Austria

**To cite this Article** Bodenteich, Michael , Faber, Kurt , Penn, Gerhard and Griengl, Herfried(1987) 'A Novel Approach to Khantiokerically Pure Carbocyclic Nucleoside Analogues', *Nucleosides, Nucleotides and Nucleic Acids*, 6: 1, 233 — 237

**To link to this Article:** DOI: 10.1080/07328318708056195

**URL:** <http://dx.doi.org/10.1080/07328318708056195>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

**A NOVEL APPROACH TO ENANTIOMERICALLY PURE  
CARBOCYCLIC NUCLEOSIDE ANALOGUES**

Michael Bodenteich, Kurt Faber, Gerhard Penn and Herfried Griengl\*

Institute of Organic Chemistry, Technical University Graz,  
Stremayrgasse 16, A-8010 Graz, Austria.

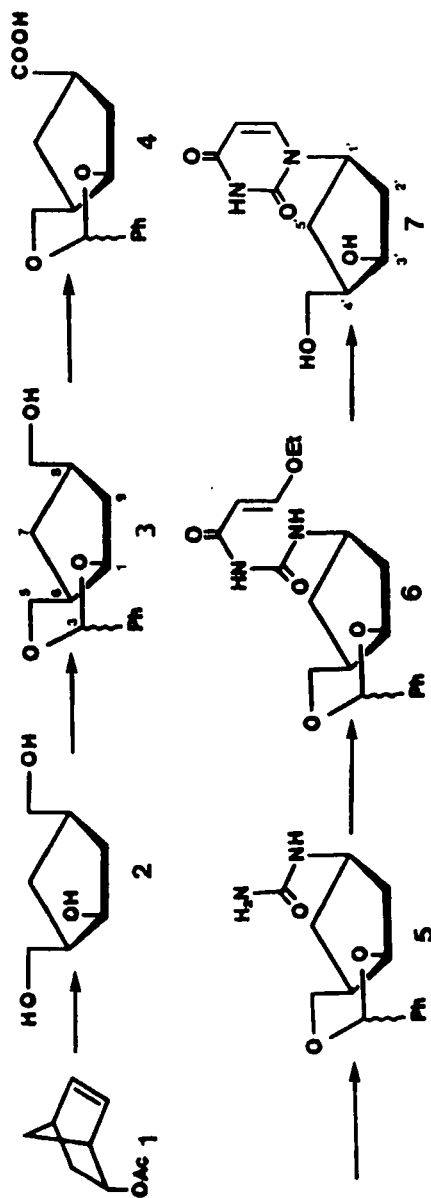
**Abstract:** Starting from (+)-endo-5-norbornen-2-yl acetate (1) (-)-1-[ (1*R*,3*R*,4*R*)-3-hydroxy-4-hydroxymethylcyclopentyl]-1*H*,3*H*-pyrimidine-2,4-dione (7) was synthesized in a 6-step sequence.

Carbocyclic analogues of nucleosides have been the topic of a variety of investigations<sup>1-5</sup> in the last years. The majority of compounds synthesized were racemic and only a few enantiomerically pure carbocyclic nucleoside analogues have been obtained either by chemoenzymatic approach<sup>2</sup>, by multistep synthesis<sup>6</sup> or by enzymatic resolution of the monophosphate of racemic aristeromycin<sup>4</sup>. As expected, in the case of aristeromycin only the (-)-enantiomer, mimicking the natural configuration of adenosine, showed antiviral activity while the corresponding (+)-enantiomer was completely inactive.

Making use of our recently published enzymatic resolution of racemic 1<sup>7</sup> we developed a short and efficient synthesis leading to carbocyclic nucleoside analogues of (-)-configuration, being equivalent to the natural occurring nucleosides. The synthetic sequence as shown in scheme 1 was published for the racemic compound in a preliminary communication<sup>8</sup>.

We want to report here the application of this approach to the synthesis of (-)-1-[ (1*R*,3*R*,4*R*)-3-hydroxy-4-hydroxymethylcyclopentyl]-1*H*,3*H*-pyrimidine-2,4-dione (7): Ozonolysis of (+)-1<sup>7</sup> (enantiomeric purity 82%) followed by reductive workup<sup>9</sup> using LiAlH<sub>4</sub> furnished triol 2 in 91% yield. Selective protection via transacetalisation with benzaldehyde dimethyl-acetal/HBF<sub>4</sub><sup>10</sup> gave quantitatively the benzylidene acetal 3. Oxidation of this material using pyridinium dichromate (PDC) in DMF<sup>11</sup> led to the

SCHEME 1



carboxylic acid **4** as a sirup (yield >90%). A neat sample of this acid crystallized spontaneously after some days of storage showing an enhanced optical rotation, which corresponds to >98% optical purity. Transformation of the sirupy material (82% e.e.) to the urea derivative **5** was achieved by successive treatment with ethyl chloroformate and  $\text{NaN}_3$ , followed by Curtius rearrangement in benzene and trapping of the intermediate isocyanate by addition of gaseous  $\text{NH}_3$ . Acylation of **5** with ethoxyacryloyl chloride<sup>12</sup> led to intermediate **6**. Finally, cyclisation under acidic conditions and simultaneous removal of the protective group furnished **7** ( $[\alpha]_D^{20}$  -55°, 82% e.e.) which could be obtained optically pure by a single recrystallisation ( $[\alpha]_D^{20}$  -67°).

### Experimental Section

Melting points were determined on a Tottoli apparatus (Büchi) and are uncorrected. Analytical TLC was performed on Merck 60 F254 silica gel plates. Medium pressure column chromatography was carried out on silica gel Merck 60, 230-400 mesh.  $^1\text{H-NMR}$  (90 MHz) and  $^{13}\text{C-NMR}$  (23 MHz) spectra were recorded on a Bruker WM-90 spectrometer with TMS as internal standard, chemical shifts ( $\delta$ ) are reported in ppm. Optical rotation values were determined on a Perkin-Elmer 141 polarimeter. All solvents were dried and distilled before use.

(-)-(1*R*,3*R*,4*R*)-4-Hydroxycyclopentane-1,3-dimethanol (**2**).

Through a solution of (+)-*endo*-5-norbornen-2-yl acetate (**1**) (7.7g, 51mmol, 82% e.e.) in 400 ml of methanol at -70° ozone was bubbled until the mixture turned blue (about 2h). Excess ozone was removed by a stream of dry  $\text{N}_2$ . The solvent was evaporated *in vacuo* at room temperature, last traces were removed at <0.1mbar. The remaining foam was dissolved in THF (100ml) and added dropwise with stirring to a cooled (-30°) suspension of  $\text{LiAlH}_4$  (7.7g, 200mmol) in 400ml of THF. When the reaction ceased the mixture was heated at reflux for 1h. Quenching was accomplished by cautious addition of saturated aqueous  $\text{MgSO}_4$ -solution (29ml) at -10° and stirring was continued for 24h. The precipitate was filtered and extracted with boiling THF for 30min. Evaporation of the combined organic phase yielded 6.43g of **2** (91%) as a colorless oil. An analytical sample was obtained by chromatography ( $\text{CHCl}_3/\text{MeOH}$  9:1 v/v);  $[\alpha]_D^{20}$  -26° (c 4.95, MeOH).  $^1\text{H-NMR}$  ( $\text{DMSO}-d_6$ ): 0.9-1.9(m, 6H, H-1, H-2, H-3, H-5), 2.94(d, J=5Hz, 2H,  $\text{CH}_2\text{OH}$ ), 3.11(dd, J=10 and 4Hz, 2H,  $\text{CH}_2\text{OH}$ ), 3.3-3.9(m, 4H, H-4, OH).  $^{13}\text{C-NMR}$  ( $\text{DMSO}-d_6$ ): 29.9, 38.2, 39.0, 47.7, 61.0, 66.1, 71.6.

(-)-(1*R*,6*R*,8*R*)-3-Phenyl-2,4-dioxabicyclo[4.3.0]nonane-8-methanol (**3**).

To a solution of triol **2** (6.1g, 42mmol) in 70ml of DMF benzaldehyde dimethylacetal (7.4g, 7.3ml, 48mmol) and  $\text{HBF}_4$  (0.3ml, 54% in ether) were added successively. The reaction was monitored by TLC (toluene/ethyl acetate 4:1 v/v) and quenched after completion (about 2h) by addition of  $\text{NEt}_3$  (1ml). The solvents were evaporated and the residue was partitioned between ether/water. After reextraction of the aqueous phase the combined organic phase was dried ( $\text{Na}_2\text{SO}_4$ ) and taken to dryness. The remaining oil was coevaporated twice with toluene and last traces of solvent were removed *in vacuo* (<0.1mbar) at 50° to give 9.7g of **3** (99%) as an oil. An analytical sample was obtained by chromatography (toluene/ethyl acetate 4:1 v/v).  $[\alpha]_D^{20}$  -17° (c 6.0, MeOH).  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ): 1.6-2.25(m, 6H), 2.8(s

br, 1H, OH), 3.61(d, J=6Hz, 2H, H-5), 4.13(s br, 2H, CH<sub>2</sub>OH), 4.2-4.35(m, 1H, H-1), 5.43(s, 1H, H-3), 7.15-7.4(m, 5H, ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 28.5, 36.5, 39.5, 40.0, 67.2, 67.4, 80.0, 100.9, 125.4, 126.2, 128.4, 129.0, 138.0.

(-)-(1*R*,6*R*,8*R*)-3-Phenyl-2,4-dioxabicyclo[4.3.0]nonane-8-carboxylic acid (**4**).

PDC (22.5g, 60mmol, 3.5eq.) was added to a solution of **3** (4.0g, 17mmol) in 30ml of DMF. After stirring for 5h (monitored by TLC, CHCl<sub>3</sub>/MeOH 9:1 v/v) the mixture was poured into 300ml of water and extracted repeatedly with ether. The organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give 3.9g of **4** (93%). An analytical sample was prepared by chromatography (CHCl<sub>3</sub>/MeOH 9:1 v/v). [ $\alpha$ ]<sub>D</sub><sup>20</sup> -10° (c 4.9, MeOH), 82% e.e. Prolonged storage of the latter led to spontaneous crystallisation yielding material which showed enhanced optical rotation [ $\alpha$ ]<sub>D</sub><sup>20</sup> -12.7° (c 5.4, MeOH), >98% e.e., mp 104-6°. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>): 1.9-2.8(m, 6H), 4.1(s br, 2H, H-5), 4.3(s br, 1H, H-1), 5.4(s, 1H, H-3), 7.15(s, 5H, ar-H), 11.5(s br, 1H, COOH).

(-)-N-[(1*R*,6*R*,8*R*)-3-Phenyl-2,4-dioxabicyclo[4.3.0]non-8-yl] urea (**5**).

To a stirred solution of **4** (3.77g, 15mmol) in acetone (80ml) was added NEt<sub>3</sub> (2.34ml, 18mmol) and ethyl chloroformate (1.63ml, 18mmol) at -70° successively. The mixture was allowed to reach -40° where it was kept for 30min and then treated with saturated aqueous NaN<sub>3</sub> solution (1.4g, 24mmol) at -10° for 30min. It was then poured into cold brine and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give an oily residue which was dissolved in benzene and heated at reflux for 30min followed by treatment with gaseous NH<sub>3</sub>. Column chromatography (CHCl<sub>3</sub>/MeOH 9:1 v/v) furnished 2.3g of the crystalline urea derivative **5** (58%). Attempts to enhance the optical purity by recrystallisation from 2-propanol were unsuccessful, mp 166-8°, [ $\alpha$ ]<sub>D</sub><sup>20</sup> -15° (c 4.2, MeOH), <sup>1</sup>H-NMR (acetone-d<sub>6</sub>): 1.2-2.5(m, 6H), 3.8-4.3(m, 4H, H-1, H-5, H-8), 5.25(s br, 2H, NH<sub>2</sub>), 5.35(s, 1H, H-3), 5.84(d, J=8.5Hz, 1H, NH), 7.1-7.4(m, 5H, ar-H).

(-)-3-Ethoxy-N-(N'-(1*R*,6*R*,8*R*)-3-phenyl-2,4-dioxabicyclo[4.3.0]non-8-yl)-carbamoyl] propenamide (**6**).

To a stirred suspension of urea **5** (1.2g, 4.6mmol) in pyridine (3ml) and CHCl<sub>3</sub> (10ml) ethoxycarbonyl chloride<sup>12</sup> (0.92g, 6.9mmol) was added and stirring was continued for 24h. After partitioning between CH<sub>2</sub>Cl<sub>2</sub>/water the organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and taken to dryness. The residue was purified by column chromatography (toluene/ethyl acetate 1:1 v/v) to give 1.34g (82%) of **6** as a foam. [ $\alpha$ ]<sub>D</sub><sup>20</sup> -1.8° (c 2.9, CH<sub>2</sub>Cl<sub>2</sub>). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>): 1.25(t, J=7Hz, 3H, CH<sub>3</sub>), 1.45-2.6(m, 5H), 3.85-4.5(m, 6H, O-CH<sub>2</sub>CH<sub>3</sub>, H-1, H-5, H-8), 5.45(s, 1H, H-3), 5.50(d, J=13Hz, 1H, CO-CH=CH-), 7.1-7.6(m, 5H, ar-H), 7.60(d, J=13Hz, 1H, CO-CH=CH-), 8.8(d, J=8Hz, 1H, NH), 10.0(s, 1H, NH).

(-)-1-[(1*R*,3*R*,4*R*)-3-Hydroxy-4-hydroxymethylcyclopentyl]-1*H*,3*H*-pyrimidine-2,4-dione (**7**).

A suspension of **6** (0.32g, 0.9mmol) in 10ml of 2*N* H<sub>2</sub>SO<sub>4</sub> was heated at 80-5° for 2.5h. The pH of the resulting solution was adjusted near 7 by addition of dil. NaOH and the solvent was removed *in vacuo*. The solid residue was extracted twice with hot ethanol (20ml each) followed by column chromatography (CHCl<sub>3</sub>/MeOH 6:1 v/v) yielding 0.15g (74%) of **7**. [ $\alpha$ ]<sub>D</sub><sup>20</sup> -55° (c=1.4, MeOH) e.e.=82%. A single recrystallisation from 2-propanol raised the optical rotation to [ $\alpha$ ]<sub>D</sub><sup>20</sup> -67° (c 0.7, MeOH) e.e.>98%, mp 183-5° [lit<sup>15</sup>: 155-7° (racemic)]. <sup>1</sup>H-NMR (acetone-d<sub>6</sub>): 1.0-2.55(m, 5H, H-2', H-4', H-5'), 3.2-3.8(m, 3H, H-3', CH<sub>2</sub>OH), 4.21(m, 1H, OH), 5.1(m, 1H, H-1'), 5.65(d, J=8Hz, 1H, H-5), 7.3(m, 1H, OH), 7.98(d, J=8Hz, 1H, H-6).

#### Acknowledgement

Financial support by Fonds zur Förderung der wissenschaftlichen Forschung (project no. P6030C) is gratefully acknowledged.

## REFERENCES

1. For leading references see: V.E.Marquez and M.-I.Lin, *Med.Res.Rev.* **6**, 1 (1986).
2. M.Arita, K.Adachi, Y.Ito, H.Sawai and M.Ohno, *J.Am.Chem.Soc.* **105**, 4049 (1983).
3. R.C.Cookson, P.J.Dudfield and D.I.C.Scopes, *J.Chem.Soc. Perkin Trans. I*, **1986**, 399.
4. P.Herdewijn, J.Balzarini, E.De Clercq and H.Vanderhaege, *J.Med.Chem.* **28**, 1385 (1985).
5. L.J.J. Hronowski and W.A. Szarek, *Can.J.Chem.* **63**, 2787 (1985).
6. C.K.H. Tseng and V.E.Marquez, *Tetrahedron Lett.* **1985**, 3669.
7. G.Eichberger, G.Penn, K.Faber and H.Griengl, *Tetrahedron Lett.* **1986**, 2843.
8. M.Bodenteich and H.Griengl, *Tetrahedron Lett.* **1986**, 4291.
9. E.Hungerbühler and D.Seebach, *Helv.Chim.Acta.* **64**, 687 (1981).
10. R.Albert, K.Dax, P.Pleschko and A.E.Stütz, *Carbohydr.Res.* **137**, 282 (1985).
11. E.J.Corey and G.Schmidt, *Tetrahedron Lett.* **1979**, 399.
12. G.Shaw and R.W.Warrener, *J.Chem.Soc.* **1958**, 153.